

Au-In (Gold-Indium)

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Figure 1 shows the Au-In phase diagram obtained by [2003Liu] by thermodynamic optimization.

The Au-In phase diagram in [Massalski2] was adopted from the evaluation done by [1987Oka]. [1992Ans] calculated the Au-In phase diagram. The result showed that the α_1 phase decomposes eutectoidally into (Au) and ζ at about 200 °C. Phases other than (Au), α_1 , and ζ were treated as line compounds in [1992Ans], although γ and ψ showed substantial solubility ranges. Accordingly, [1993Oka] introduced the work of [1992Ans] by modifying the diagram of [1987Oka] in the region involving (Au), α_1 , and ζ according to [1992Ans] (Fig. 2). Because the solubility of In in (Au) must become 0 at.% at 0 K, the (Au) solvus calculated by [2002Ans] appears to require unlikely inflection when extrapolated below 0 °C. Accordingly, the work of [2003Liu], which shows more similarity to [1987Oka], may be a better

representation of the true equilibrium. However, this result needs experimental corroboration.

References

- 1987Oka:** H. Okamoto and T.B. Massalski: "The Au-In (Gold-Indium) System" in *Phase Diagrams of Binary Gold Alloys*, H. Okamoto and T.B. Massalski, ed., ASM International, Metals Park, OH, 1997, pp. 142-53.
- 1992Ans:** I. Ansara and J.P. Nabot: "A Thermodynamic Re-Assessment of the Au-In System in the Au-rich Region," *Calphad*, 1992, 16(1), pp. 13-18.
- 1993Oka:** H. Okamoto: "Au-In (Gold-Indium)," *J. Phase Equilibria*, 1993, 14(4), pp. 532-33.
- 2003Liu:** H.S. Liu, Y. Cui, K. Ishida, and Z.P. Jin: "Thermodynamic Reassessment of the Au-In Binary System," *Calphad*, 2003, 27(1), pp. 27-37.

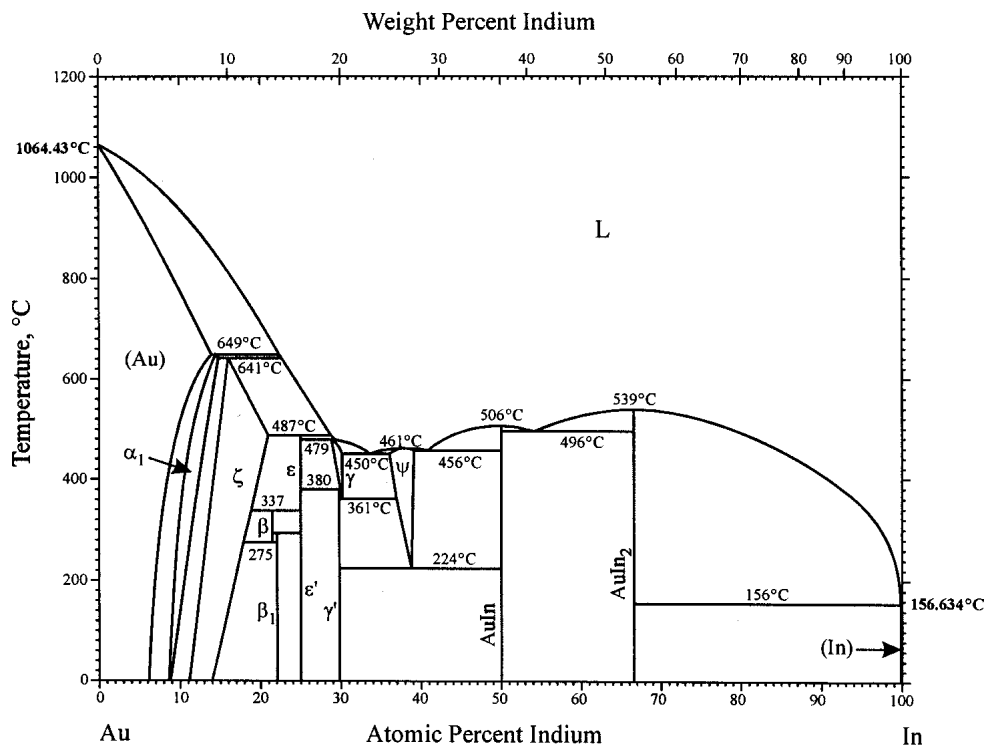


Fig. 1 Au-In phase diagram calculated by [2003Liu]

Section III: Supplemental Literature Review

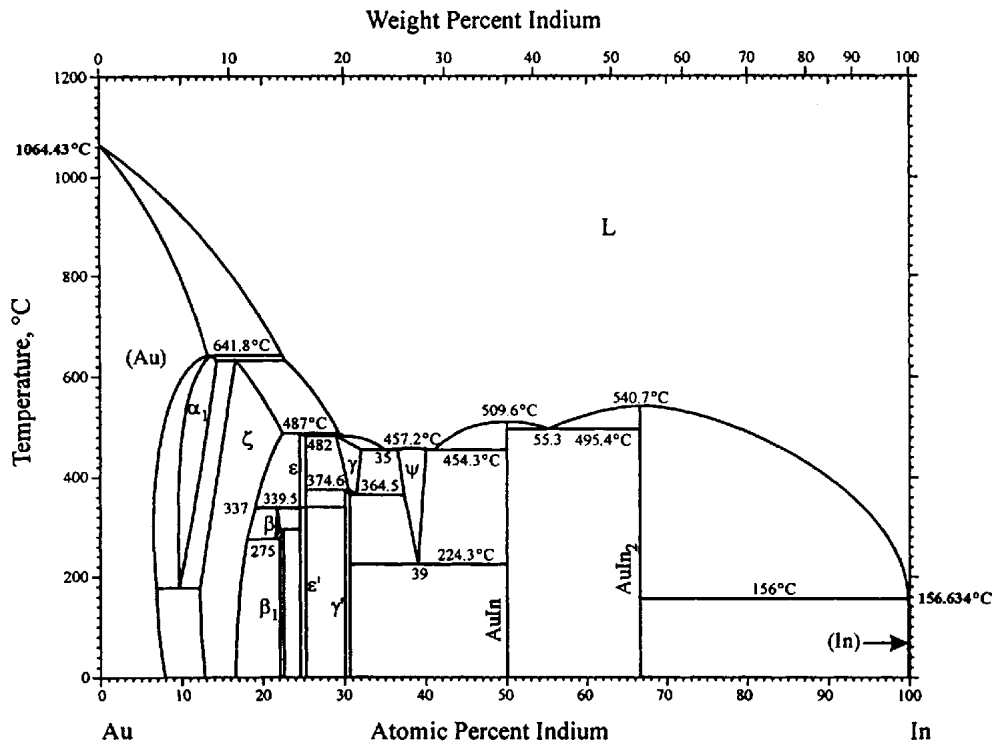


Fig. 2 Au-In phase diagram in [1993Oka]. The region involving (Au), α_1 , and ζ is from [1992Ans] and the rest is from [1987Oka].